

HYDROGEN PEROXIDE (H₂O₂) (Perhydrol, Hydroxyperoxide, Oxydol)

PRODUCT SPECIFICATIONS:

		% 30	%35	%50	%60	%70
Concentration (% m/m)	:	≥30	≥35	≥30	≥30	≥30
Acidity (% H ₂ SO ₄)	:	≤0,04	≤0,04	≤0,04	≤0,045	≤0,045
Stability (%m/m)	:	≥97	≥97	≥97	≥97	≥97
Appearance	:	Clear, colorless liquid				

PHYSICAL AND CHEMICAL PROPERTIES:

Concentration	% 30	%35	%50	%60	%70
Boiling Point (°C)	106	108	114	119	126
Freezing Point (°C)	-25	-32	-52	-56	-40
Active O ₂ Content (%)	14,1	16,4	23,5	28,5	32,8
Relative density (20°C/4°C)	1,11	1,13	1,19	1,24	1,29
Total Vapour Pressure (30°C) torr	25	24	18	14	11
Partial Vapour Press (30°C) torr	0,25	0,3	0,6	0,9	1,3
	Boiling Point (°C) Freezing Point (°C) Active O ₂ Content (%) Relative density (20°C/ 4°C) Total Vapour Pressure (30°C) torr	Boiling Point (°C) 106 Freezing Point (°C) -25 Active O_2 Content (%) 14,1 Relative density (20°C/ 4°C) 1,11 Total Vapour Pressure (30°C) torr 25	Boiling Point (°C) 106 108 Freezing Point (°C) -25 -32 Active O_2 Content (%) 14,1 16,4 Relative density (20°C/4°C) 1,11 1,13 Total Vapour Pressure (30°C) torr 25 24	Boiling Point (°C) 106 108 114 Freezing Point (°C) -25 -32 -52 Active O_2 Content (%) 14,1 16,4 23,5 Relative density (20°C/4°C) 1,11 1,13 1,19 Total Vapour Pressure (30°C) torr 25 24 18	Boiling Point (°C) 106 108 114 119 Freezing Point (°C) -25 -32 -52 -56 Active O2 Content (%) 14,1 16,4 23,5 28,5 Relative density (20°C/ 4°C) 1,11 1,13 1,19 1,24 Total Vapour Pressure (30°C) torr 25 24 18 14

PACKAGING:

Delivered in 65 kg plastic drums, 1 m³ IBC tanks and bulk in ISO containers.

ANALYTICAL METHOD:

Method : Titrimetric

Reagents : 0,1 N KMnO₄ and % 5 H₂SO₄ (Analytical purity)

Precedure : Take 0,15-0,20 gr of sample, put weighing bottle in a 250 mL erlenmayer flask containing 100 mL

of %5 sulfiric acid, titrate with 0,1 N potassium permanganate standart solution to a pink color, which does not

dissappears in 30 sec.

Calculation:

F x V x 0,17 % H₂O₂ = -----

F: Factor of 0,1 N KMnO₄ Standart solution V: Volume of 0,1 N KMnO₄ standart solution (mL) G: weight of hydrogen peroxide sample, (gr)

APPLICATION FIELDS

Industry of Textile
 Industry of Pulp and Paper
 As a bleaching agent.
 As a bleaching agent.

• Industry of Chemistry : In oxidation and hydroxilation reaction; also in the production of

organic/inorganic peroxy compound like perasetic acid, sodium perborate, sodium percarbonate and calcium peroxide.

• Industry of Environmental Chemicals: Waste water treatment; as a detoxifying agent and provides dissolved

oxygen.

Food Processing
 Industry of Pharmaceutical
 Industry of Cosmetic
 Industry of Mining
 Sterilization of packaging of milk, fruit, juices etc.
 Topical antiseptic and contact lens cleaner.
 Hair bleaching and dye setting through oxidation.
 Removing poisonous effects of different mines.

Industry of Metallurgy
 Industry of Pool Chemical
 Water cleaning and recovery.





STORAGE AND HANDLING:

- Hydrogen Peroxide should be stored in a cool place, away from direct sunlight, heat and oxidizing agents.
- Keep the empty containers sealed to prevent the penetration of any catalytically active contaminant.
- Avoid the contact with organic material or oxidizable product. Even though the product is not flammable itself, the concentrated solution will act as a strong oxidizing agent and may ignite flammable materials.
- Always keep sufficient water to wash the area, in case of an accidental leakage or spill. Aerate the place in order to remove the hydrogen peroxide vapor.
- Wear a protective clothing, to avoid the corrosive effect of hydrogen peroxide on skin, mucous membraines and respiratory passages.
- Materials that are suitable for storage of hydrogen peroxide:

Stainless steel (304/316 L)

Pure aluminium (min % 99,5)

Tantalum

Zirconium

Glass / Ceramic

Polyethylene and PVC (They can only be used for the concentrations up to 60 wt %)

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